

Synthesis of a tin compound bearing N-alkoxy carboxamide and methyl ligands as a precursor for SnO₂ fabrication via atomic layer deposition

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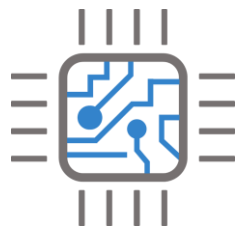
I Overview of Our R&D Team

II Experiment background

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Overview of Our R&D Team



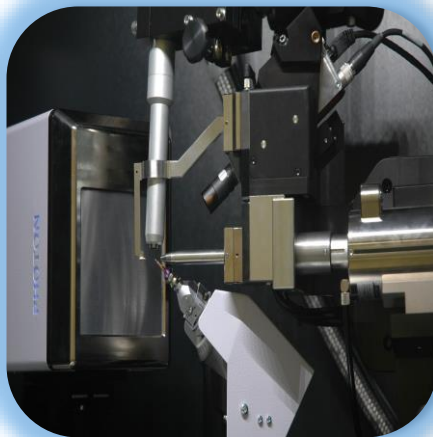
Development of Core Technologies for Advanced Electronic and Information Materials

Molecular
Design
ALD/CVD
Precursors

Precursors
characterization

Development of
deposition
process

Applications in
Semiconductors



Experiment background

■ Application of oxide semiconductor

Display



Optoelectronic



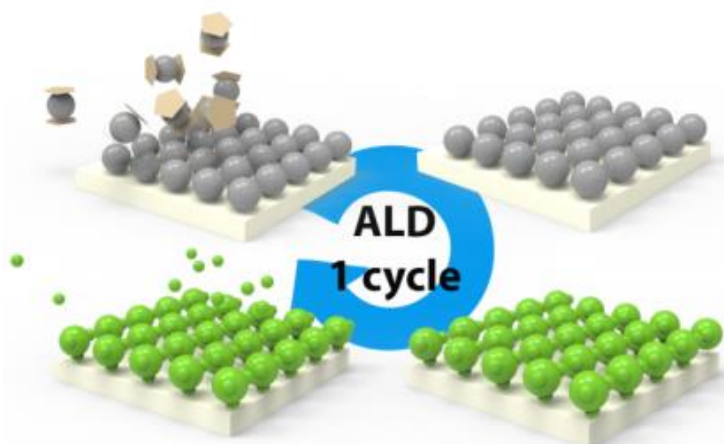
Sensor



Battery



■ Mechanism of atomic layer deposition (ALD)



1. Sequential surface chemical reaction process

- Each precursor supplying **sequentially** without thermal decomposing
- Forming films by **repeating cycles**

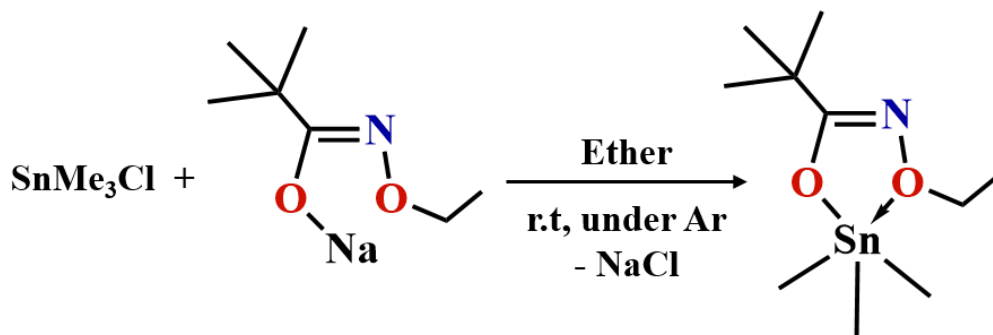
2. Chemisorption saturation process

- Precursor vapor distributing on surface and saturated by self-limited reaction
- **Surface reaction** and removing physisorbed precursor by inert gas
 - **Exact thickness control in nano-scale**
 - **Excellent large area uniformity**
 - **Superior step coverage contrast to PVD and CVD**

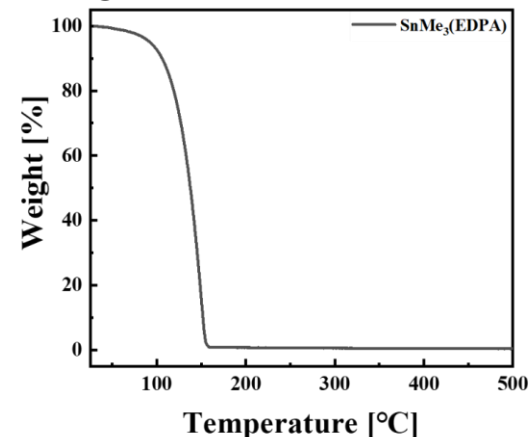
Result

Precursor (Synthesis and characteristic)

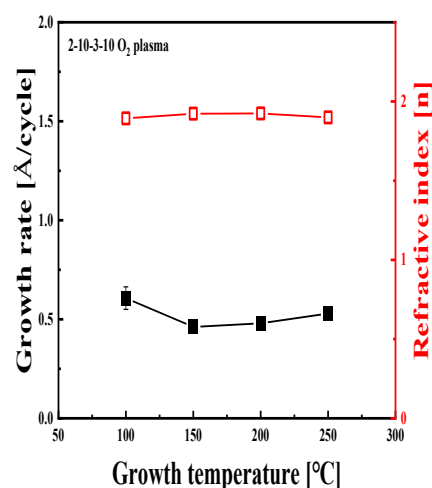
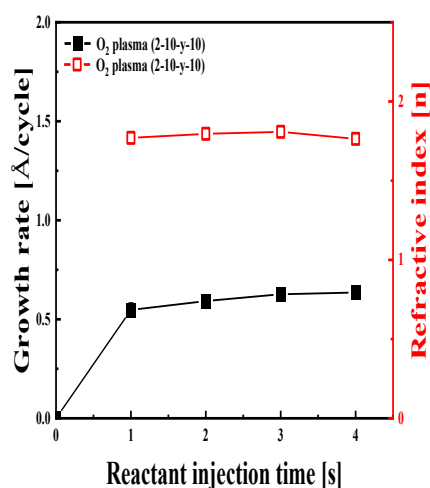
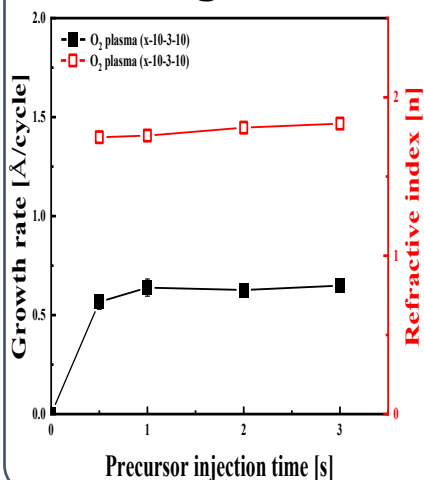
Synthesis



TGA



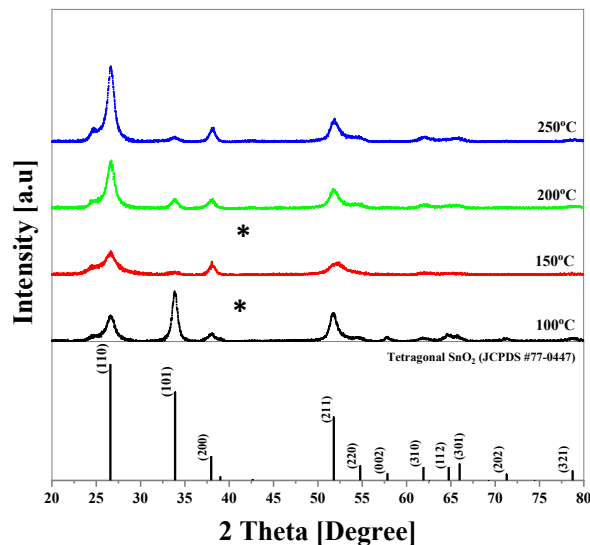
ALD growth behavior



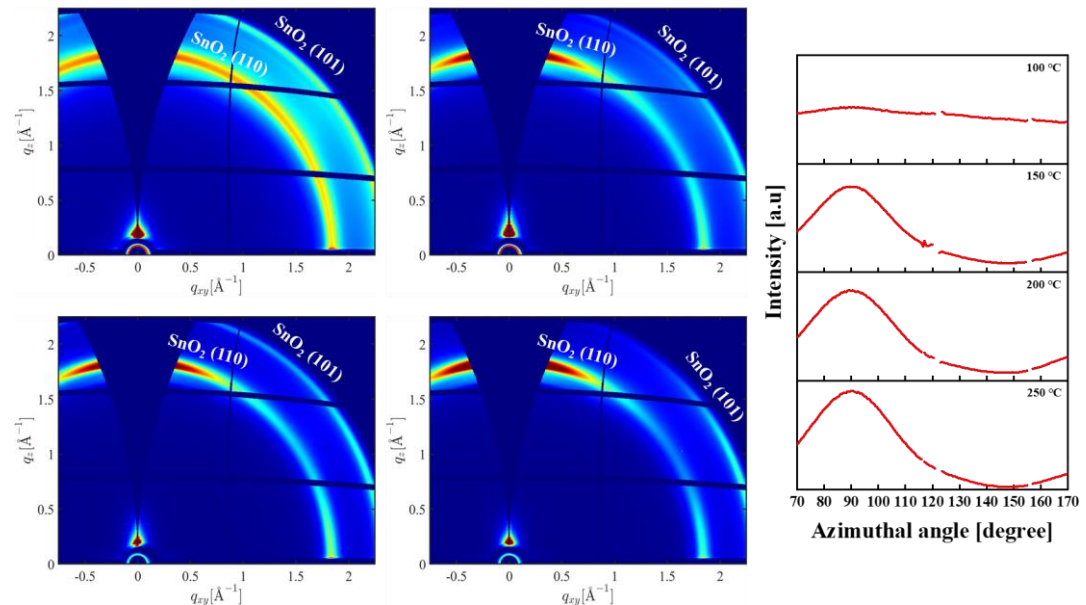
- SnO₂ film exhibited typical self-limited surface reaction characteristics
- ALD window in the growth temperature range between 100 – 250 °C

Result

Structural analyses



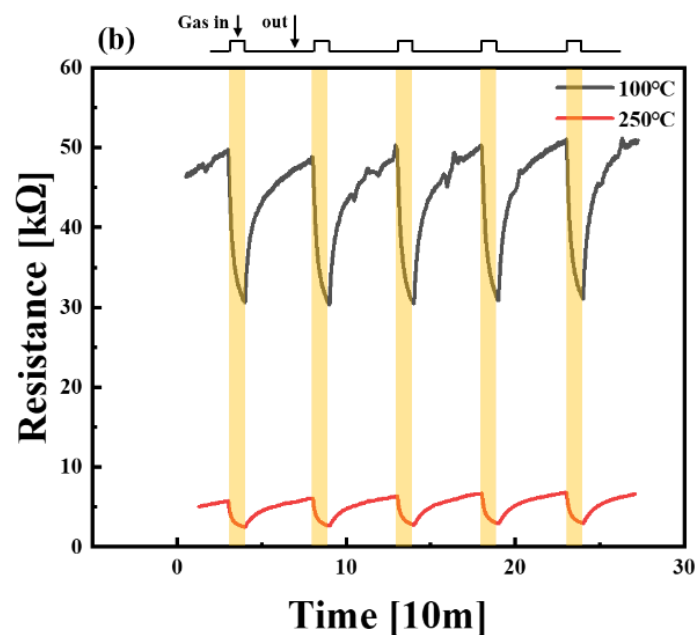
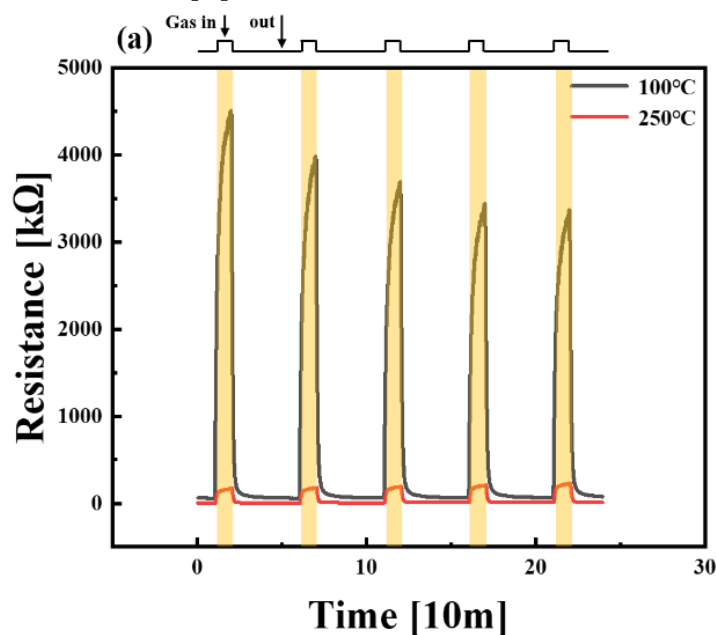
- At 100 – 250 °C, tetragonal SnO₂ formed
- Temperature increased, the intensity of (110) plane increased



- At all deposition temperatures, the X-ray reflections are observed at values of 1.875 and 2.376 Å⁻¹, which indicate the (110) and (101) planes of the tetragonal SnO₂
- (110) plane of the SnO₂ thin films deposited at 150–250 °C indicate a preferential growth to the c-axis, whereas the (101) plane exhibits a random orientation

Result

Gas sensor application



Material used	Deposition	Gas	Temperature (°C)	Response (%)
Thin film	sprayed	NO ₂	350	60(500 ppm)
Nanobelts	Carbothermal reduction	NO ₂	200	45(50 ppm)
Thin film	Sol gel	NO ₂	Not detailed	25(500 ppm)
Thin film	CBD	NO ₂	Not detailed	44(100 ppm)
Nanotubes	electrospinning	NH ₃	Room temp	2.48(100 ppm)
Thin film	sputtering	NH ₃	230	25.7(450 ppm)
Deposited at 100 °C	ALD	NO₂	300	6000(100 ppm)
Deposited at 100 °C	ALD	NH ₃	300	40(100 ppm)
Deposited at 250 °C	ALD	NO ₂	300	2500(100 ppm)
Deposited at 250 °C	ALD	NH ₃	300	36(100 ppm)

Conclusion

- We successfully deposited tin oxide by atomic layer deposition (ALD) using a novel precursor and O₂ plasma.
- The properties of films through various analysis methods. As a result, SnO₂ films deposited at low temperatures have significant potential as high-performance gas sensor materials.